

# Rate of Diadenosine Polyphosphate Degradation by Various Guinea Pig Tissues

A. U. Ziganshin, L. E. Ziganshina, and G. Burnstock

Translated from *Byulleten' Eksperimental'noi Biologii i Meditsiny*, Vol. 127, No. 2, pp. 207-210, February, 1999  
Original article submitted April 8, 1998

High-performance liquid chromatography and colorimetry of inorganic phosphorus show that in the presence of muscle tissues from the guinea pig bladder, seminal duct, small intestine, longitudinal cord of the cecum, and right atrium all diadenosine polyphosphates with a common formula  $AP_nA$  ( $n=2-6$ ) are degraded more slowly than ATP. The rate of  $AP_nA$  cleavage is highest in experiments with the small intestinal and right atrial tissues. P2-purine receptor antagonist suramin inhibits degradation of  $AP_3A$  by right atrial tissue and does not modify degradation of other  $AP_nA$ . Enzymatic stability of endogenous  $AP_nA$  is apparently the principal condition for manifestation of their biological effects.

**Key Words:** *diadenosine polyphosphates; ectonucleotidases; P2-purine receptors; suramin*

Diadenosine polyphosphates are a family of dinucleotides consisting of two adenosine molecules bound through polyphosphate chain in the 5' position. Diadenosine polyphosphates are usually denoted as  $AP_nA$ , where  $n$  is the number of phosphoric acid residues (P), usually 1-6.  $AP_nA$  with  $n=4-6$  attract the greatest attention, because these substances were isolated from the nervous system and visceral tissues of humans and animals [3,6,9].  $AP_nA$  released from secretory cells and nerve terminals participate in the regulation of various cell processes via extracellular receptors.  $AP_nA$  are believed to be involved in the regulation of blood pressure and clotting [6,9]. The receptors activated by  $AP_nA$  are heterogeneous. These compounds interact with certain subtypes of P2-purine receptors [5,8]. In the CNS,  $AP_nA$  affect specific dinucleotide receptors different from P2-purine receptors [2].

Extracellular effects of  $AP_nA$  are probably arrested by specific ectonucleotidases cleaving the polyphosphate chain molecules into two parts and yielding adenosine tri-, di-, and monophosphates, which are then catabolized via usual metabolic pathways, i.e. by successive cleavage of phosphoric acid residues [7].

Such ectonucleotidase has been described for the adrenal chromaffine cells [1].

Therefore, if  $AP_nA$  is a mediator, ectonucleotidase is an enzyme limiting their effect. We studied the enzymatic stability of  $AP_nA$  in the presence of tissues most often used in pharmacological studies of P2-purine receptors: the guinea pig urinary bladder, seminal duct, taenia caeci, small intestine, and atrium.

## MATERIALS AND METHODS

Experiments were carried out on tissues from male guinea pigs weighing 350-600 g. The animals were euthanized and dehematized. The bladder, seminal duct, taenia caeci, a fragment of the small intestine, and heart were removed, and muscle preparations were made as for a pharmacological experiment.

The rate of  $AP_nA$  degradation was measured at  $37\pm 1^\circ\text{C}$  in a buffer containing 10 mM 4-(2-hydroxyethyl)-1-piperazine ethanesulfonic acid (HEPES), 135 mM NaCl, 5 mM KCl, 2 mM  $\text{CaCl}_2$ , 2 mM  $\text{MgCl}_2$ , and 10 mM glucose, pH 7.4. Tissue fragments (2-3 mg) were incubated in a 24-well plate in 250  $\mu\text{l}$  buffer containing individual  $AP_nA$  (or ATP) in a concentration of 100  $\mu\text{M}$  for 30 min (10 min for ATP) at constant shaking.

Department of Pharmacology, Kazan State Medical University; Institute of Autonomic Nervous System, London

In the first series, the reaction was stopped by transferring the buffer to 0.9 ml 2.5% sodium dodecyl sulfate. The concentration of formed inorganic phosphorus ( $P_i$ ) was measured [10] using the corresponding concentrations of  $K_2HPO_4$  reference solution. In the absence of substrates (control) no  $P_i$  was detected.

In the second series, APnA were assayed by high-performance liquid chromatography (HPLC). The reaction buffer was immediately frozen in liquid nitrogen. The chromatographic system included a Beckman 114M injector attached to a SA6500 UV/Vis spectrophotometer. APnA were separated on a 25-cm Spherisorb column (Hichrom 0.46 mm internal diameter). The column was washed for 24 h with a mobile phase consisting of 0.2 M  $KH_2PO_4$  with 3% methanol (pH 6.0) at a flow rate of 0.2 ml/min. The separation of APnA (20- $\mu$ l sample) was carried out at a flow rate of 1.5 ml/min (detection 260 nm wavelength). Operations with AP4A, AP5A, and AP6A were carried out using the same mobile phase as for washing the column, while for AP2A and AP3A the concentration of methanol in the mobile phase was brought to 10% to accelerate the process. The amount of APnA was evaluated by comparing the peaks of experimental samples and standard APnA solutions.

In the reference experiments ATP was used as the reaction substrate. Mean values of at least 3 experiments in 4 parallels were calculated. All reagents were from Sigma.

## RESULTS

All studied tissues metabolized APnA, the rate of APnA cleavage being far lower than of ATP. The rate of APnA degradation was the highest in samples containing the cecum and right atrium (25-30% ATP degradation rate, Table 1). The lowest rate of APnA degradation was observed in the presence of the bladder and seminal duct tissues (<5% ATP degradation rate). AP3A displayed the highest enzymatic stability in the presence of the bladder and taenia caeci tissues, while AP2A degraded most slowly in the presence of

small intestinal and right atrial tissues. In the seminal duct all APnA degraded at the same rate.

Therefore, all APnA are metabolized in the studied tissues, although to a different degree. However, the method for measuring  $P_i$  does not show the true rate of substrate degradation. For example, complete degradation of the entire polyphosphate chain in AP6A molecule yields 6  $H_3PO_4$  molecules, and hence, the rate of degradation can be 6-fold overestimated. More accurate measurements were carried out using HPLC in samples containing guinea pig atrium, characterized by the most intense APnA degradation.

As expected, the values were lower than in the first experimental series. AP3A had the highest rate of degradation in this tissue (about 16% of the rate of ATP degradation in the same tissue). Other APnA exhibited similar degradation rates (Table 2).

Bearing in mind that P2-purine receptor antagonist suramin effectively inhibits ectoATPase [11], we studied its effect on APnA degradation. Suramin significantly inhibited degradation of AP3A and ATP but not of other APnA (Table 2). This may be explained by the fact that AP3A is cleaved by ectonucleotidases similar to ectoATPases.

The physiological and pathophysiological role of APnA is still discussed. On the one hand, these substances are present in cells in very low concentrations: in the majority of studied tissues the ATP:APnA ratio is 25:1, and even 100:1 in human platelets [3]. On the other hand, these compounds are characterized by a longer half-life compared to other endogenous mediators [1,4]. The effects of APnA are mediated via P1- or P2-purine receptors and specific dinucleotide receptors and can be oppositely directed. *In vitro* AP2A, AP3A, and AP4A exert vasodilatory, while AP5A and AP6A vasoconstrictory effects. Consequently, intravenous infusion of AP4A reduces and of AP5A and AP6A elevates blood pressure [7]. Like adenosine, APnA exert cardiodepressive effects which are apparently mediated by type A1 P1-purine receptors [5]. APnA stimulate the contractility of isolated urinary bladder, and this effect is comparable to that of ATP

TABLE 1. The Rate of  $P_i$  Release from APnA in the Presence of Various Guinea Pig Tissues, pmol  $P_i$ /mg tissue/min ( $M \pm m$ )

Substrate	Urinary bladder	Seminal duct	Taenia caeci	Small intestine	Right atrium
ATP	348.7 $\pm$ 12.8 (100)	364.0 $\pm$ 13.1 (100)	303.8 $\pm$ 9.8 (100)	579.3 $\pm$ 30.1 (100)	595.5 $\pm$ 27.2 (100)
AP2A	15.1 $\pm$ 6.5 (4.3)	5.1 $\pm$ 2.2 (1.4)	21.0 $\pm$ 8.0 (6.9)	86.1 $\pm$ 5.4 (14.9)	120.5 $\pm$ 9.8 (20.2)
AP3A	4.6 $\pm$ 1.5 (1.3)	6.2 $\pm$ 2.5 (1.7)	14.9 $\pm$ 5.3 (4.9)	114.2 $\pm$ 12.2 (19.7)	184.0 $\pm$ 13.9 (30.9)
AP4A	12.2 $\pm$ 6.0 (3.5)	8.5 $\pm$ 3.9 (2.3)	32.6 $\pm$ 9.6 (10.7)	152.8 $\pm$ 14.6 (26.3)	164.5 $\pm$ 14.9 (27.6)
AP5A	10.7 $\pm$ 5.4 (3.1)	10.6 $\pm$ 3.8 (2.9)	20.7 $\pm$ 4.7 (6.8)	147.7 $\pm$ 10.1 (25.5)	163.5 $\pm$ 15.7 (27.4)
AP6A	14.4 $\pm$ 6.0 (4.1)	16.1 $\pm$ 5.4 (4.4)	22.7 $\pm$ 5.4 (7.5)	123.3 $\pm$ 11.8 (21.2)	178.7 $\pm$ 15.9 (30)

Note. Here and in Table 2: the percentage of ATP degradation rates is given in parentheses.

**TABLE 2.** Rate of AP<sub>n</sub>A Degradation in Experiments with Guinea Pig Right Atrial Tissue with and without Suramin, pmol AP<sub>n</sub>A/mg tissue/min ( $M \pm m$ )

Substrate	No suramin	Suramin, 300 $\mu$ M
ATP	341.5 $\pm$ 10.7 (100)	175.5 $\pm$ 6.8* (100)
AP2A	11.8 $\pm$ 3.9 (3.4)	8.5 $\pm$ 4.5 (2.5)
AP3A	52.5 $\pm$ 4.3 (15.4)	17.5 $\pm$ 2.6* (5.1)
AP4A	13.0 $\pm$ 2.0 (3.8)	6.0 $\pm$ 3.0 (1.8)
AP5A	18.5 $\pm$ 3.9 (5.4)	12.5 $\pm$ 2.5 (3.7)
AP6A	13.6 $\pm$ 0.9 (3.9)	8.3 $\pm$ 1.1 (2.4)

Note. \* $p < 0.05$  vs. samples without suramin.

and mediated by P2X-purine receptors [8]. Such a variety of effects may be explained by modification of the AP<sub>n</sub>A molecule due to enzymatic degradation, depending on the receptor type.

Therefore, AP<sub>n</sub>A are far superior to ATP in their enzymatic stability in all tissues studied. The content of AP<sub>n</sub>A in tissues is relatively low, but unlike ATP which is rapidly destroyed by ectoATPases, AP<sub>n</sub>A are characterized by longer half-life in tissues and a prolonged effect on cell receptors even if released in low

amounts, which probably determines the biological role of these molecules.

## REFERENCES

1. A. Ogilvie, J. Luthje, U. Pohl, *et al.*, *Biochem. J.*, **259**, 97-103 (1989).
2. J. Pintor, J. Gualix, and M. T. Miras-Portugal, *Mol. Pharmacol.*, **51**, 277-284 (1997).
3. J. Pintor, C. H. V. Hoyle, J. Gualix, *et al.*, *Neurosci. Res.*, **20**, 69-78 (1997).
4. J. Pintor and M. T. Miras-Portugal, *Gen. Pharmacol.*, **26**, 229-235 (1995).
5. A. Rubino and G. Burnstock, *J. Physiol.*, **495**, No. 2. 515-523 (1996).
6. H. Schluter, E. Offers, G. Bruggemann, *et al.*, *Nature*, **367**, 186-188 (1994).
7. H. Schluter, M. Tepel, and W. Zidek, *J. Auton. Pharmacol.*, **16**, 357-362 (1996).
8. T. D. Westfall, C. A. McIntyre, S. Obeid, *et al.*, *Brit. J. Pharmacol.*, **121**, 57-62 (1997).
9. P. C. Zamecnik, B. Kim, M. J. Gao, *et al.*, *Proc. Natl. Acad. Sci. USA*, **89**, 2370-2373 (1992).
10. A. U. Ziganshin, L. E. Ziganshina, C. H. V. Hoyle, *et al.*, *Brit. J. Pharmacol.*, **114**, 632-639 (1995).
11. A. U. Ziganshin, L. E. Ziganshina, B. F. King, *et al.*, *Biochem. Pharmacol.*, **51**, 897-901 (1996).